

Performance of Kier-Hall E-states descriptors in QSAR of multi-functional molecules

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Kier-Hall E-state descriptors

- Pharm.Res. **1990**, 7, 801-807
- JCICS 1991, 31, 76-82
- $I = (\delta^V + 1) / \delta$
 - δ^V and δ are counts of valence and sigma electrons of atoms associated with the molecular skeleton
- $S_i = I_i + \Delta I_i$
 - E-state value, S_i , for skeletal atom i
- ΔI_i , is given as $\sum (I_i - I_j) / r_{ij}^2$

Intrinsic-State Values

J. Chem. Inf. Comput. Sci., Vol. 31, No. 1, 1991 77

Table I. Intrinsic-State Values

atom (skeletal hydride group)	intrinsic-state value ^a
)C(1.250
)CH-	1.333
-CH ₂ -	1.500
)C=	1.667
-CH ₃ , =CH-,)N-	2.000
≡C-, -NH-	2.500
=CH ₂ , =N-	3.000
-O-	3.500
≡CH, -NH ₂	4.000
=NH	5.000
≡N, -OH	6.000
=O	7.000
-F	8.000
-Cl	4.111
-Br	2.750
-I	2.120
=S	3.667
-SH	3.222
-S-	1.833

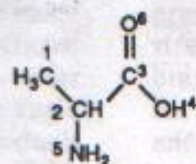
^aCalculated from eq 2.

Kier-Hall Atom Types

RowNo	atom-types-Kier-Hal	RowNo	atom-types-Kier-Hal
1	sOH	19	ddssS
2	dO	20	sF
3	ssO	21	sCl
4	aaO	22	sBr
5	sNH2	23	sl
6	dNH	24	sCH3
7	ssNH	25	ssCH2
8	aaNH	26	dCH2
9	tN	27	sssCH1
10	dsN	28	dsCH1
11	aaN	29	tCH
12	sssN	30	aaCH
13	ddsN	31	aasC
14	ssssN+	32	ddC
15	sSH	33	tsC
16	dS	34	dssC
17	ssS	35	ssssC
18	aaS		

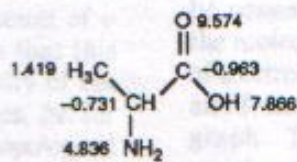
Kier-Hall Algorithm

Table II. Electrotopological-State Calculations for Alanine



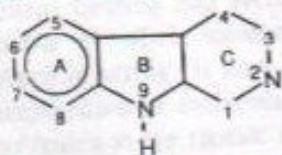
intrinsic values		intrinsic values					intrinsic values	
$I(1) = 2.000$		$I(3) = 1.667$					$I(5) = 4.000$	
$I(2) = 1.333$		$I(4) = 6.000$					$I(6) = 7.000$	
$(I_i - I_j)/r_{ij}^2$ Matrix								
i	1	2	3	4	5	6	$\Delta I =$ row sum	
1	0.0	0.1667	0.0370	-0.2500	-0.2222	-0.3125	-0.5810	
2	-0.1667	0.0	-0.083	-0.5185	-0.6667	-0.6296	-2.0648	
3	-0.0370	0.0833	0.0	-1.0833	-0.2593	-1.3333	-2.6296	
4	0.2500	0.5185	1.0833	0.0	0.1250	-0.1111	1.8657	
5	0.2222	0.6667	0.2593	-0.1250	0.0	-0.1875	0.8356	
6	0.3125	0.6296	1.3333	0.1111	0.1875	0.0	2.5741	
							0.0000	

$$S_i = I_i + \Delta I_i$$



QSAR example 1

Table 6. Binding of beta-carbolines to the benzodiazepine receptor and electrotopological state values

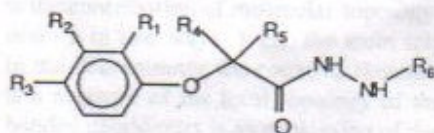


Obs	R ₁	R ₃	R ₄ ^a	R ₅	R ₆	R ₇	S(CO) ^b	S(NN)	S(C ₁₄)	pIC ₅₀ ^c	Calc ^d	Res ^e
1	CH ₃	H	H	oH	H	H	0.930	3.828	1.555	-0.450	-1.166	0.716
2	CH ₃	H	H	H	H	OCH ₃	0.921	3.831	1.523	-1.980	-1.228	-0.752
3	CH ₃	H	H	H	H	OH	0.902	3.748	1.480	-1.900	-2.505	0.605
4	H	CH ₂ OH	H	H	H	H	4.867	3.712	1.845	1.591	1.972	-0.381
5	H	COOCH ₃	H	H	H	H	5.872	3.658	1.702	2.097	1.698	0.399
6	H	COOC ₂ H ₅	H	H	H	H	5.997	3.688	1.719	2.155	2.277	-0.122
7	H	COOC ₂ H ₇	H	H	H	H	6.082	3.711	1.731	1.921	2.702	-0.781
8	H	COOCH ₃	C ₂ H ₅	H	OCH ₃	OCH ₃	6.162	3.787	1.253	2.400	2.146	0.254
9	H	COOC ₂ H ₅	CH ₂ OCH ₃	OCH ₂ C ₆ H ₅	H	H	6.360	3.850	1.170	2.959	2.875	0.084
10	H	COOC ₂ H ₅	CH ₃	O-C ₃ H ₇	H	H	6.225	3.800	1.236	2.960	2.319	0.641
11	H	COONHCH ₃	H	H	H	H	5.934	3.661	1.685	1.780	1.737	0.043
12	CH ₃	H ₂	H ₂	H	H	OCH ₃	0.451	3.947	1.068	-2.813	-1.708	-1.105
13	CH ₃	H ₂	H ₂	H	H	OH	0.434	3.864	1.023	-2.748	-2.969	0.221
14	H	COOC ₂ H ₅	CH ₂ OCH ₃	H	OCH ₂ C ₆ H ₅	H	6.335	3.836	1.181	2.959	2.703	0.256
15	H	H	H	H	H	H	0.911	3.696	1.946	-1.000	-1.587	0.587
16	H	H	OOCH	H	H	H	5.448	3.530	1.574	-1.491	-0.825	-0.666

- a. In compounds 12 and 13 the C ring is not aromatic; all others are.
 b. $S(\text{CO}) = (S(\text{C}_3) + S(\text{O}))/2$; $S(\text{O})$ is the E-state value for the oxygen attached to the carbon atom in the substituent on position 3. See text.
 $S(\text{NN}) = (S(\text{N}_2) + S(\text{N}_9))/2$. See text. $S(\text{C}_{14}) = (S(\text{C}_1) + S(\text{C}_4))/2$. See text.
 c. The negative log of the IC₅₀ value for binding.
 d. Calc is the value computed for pIC₅₀ from Eq. 9.
 e. Res = Calc - pIC₅₀.

QSAR example 2

Table 5. Hydrazide monoamine oxidase inhibitors and their electrotopological state values



Obs	R ₁	R ₂	R ₃	R ₄	R ₅	R ₆	S(CH ₁₄) ^a	S(NH ₁₁) ^b	pIC ₅₀ ^c	Calc ^d	Res ^e
1	H	H	H	H	H	CH(CH ₃) ₂	1.943	2.538	5.42	5.42	0.00
2	Cl	H	H	H	H	CH(CH ₃) ₂	1.926	2.515	5.60	5.66	-0.06
3	H	Cl	H	H	H	CH(CH ₃) ₂	1.929	2.520	5.40	5.61	-0.21
4	H	H	Cl	H	H	CH(CH ₃) ₂	1.931	2.524	5.96	5.57	0.39
5	CH ₃	H	H	H	H	CH(CH ₃) ₂	1.947	2.558	5.54	5.22	0.32
6	H	CH ₃	H	H	H	CH(CH ₃) ₂	1.946	2.553	5.05	5.27	-0.22
7	H	H	CH ₃	H	H	CH(CH ₃) ₂	1.946	2.550	5.40	5.30	0.10
8	OCH ₃	H	H	H	H	CH(CH ₃) ₂	1.932	2.535	5.62	5.47	0.15
9	H	OCH ₃	H	H	H	CH(CH ₃) ₂	1.934	2.536	5.42	5.45	-0.03
10	H	H	OCH ₃	H	H	CH(CH ₃) ₂	1.935	2.536	5.52	5.45	0.07
11	H	H	H	H	CH ₃	CH(CH ₃) ₂	1.947	2.588	5.00	4.94	0.06
12	Cl	H	H	H	CH ₃	CH(CH ₃) ₂	1.930	2.565	5.16	5.19	-0.03
13	H	Cl	H	H	CH ₃	CH(CH ₃) ₂	1.933	2.570	4.96	5.13	-0.17
14	H	H	Cl	H	CH ₃	CH(CH ₃) ₂	1.936	2.573	5.00	5.10	-0.10
15	H	H	H	CH ₃	CH ₃	CH(CH ₃) ₂	1.950	2.629	4.34	4.55	-0.21
16	H	H	Cl	CH ₃	CH ₃	CH(CH ₃) ₂	1.938	2.614	4.80	4.71	0.09
17	H	CH ₃	H	H	CH ₃	CH(CH ₃) ₂	1.951	2.603	4.90	4.79	0.11
18	H	H	H	H	H	C ₂ H ₅	1.901	2.489	5.82	5.95	-0.13
19	H	H	Cl	H	H	C ₂ H ₅	1.889	2.474	6.00	6.12	-0.12
20	H	H	H	H	H	CH ₂ C ₆ H ₅	1.103	2.587	6.14	6.44	-0.30
21	H	H	H	H	H	CH(CH ₃)C ₆ H ₅	1.108	2.637	5.70	5.96	-0.26
22	H	H	CH ₃	H	H	CH(CH ₃)C ₆ H ₅	1.109	2.649	6.05	5.85	0.20
23	H	H	OCH ₃	H	H	CH(CH ₃)C ₆ H ₅	1.096	2.635	6.00	6.00	-0.00
24	H	H	Cl	H	H	CH ₂ C ₆ H ₅	1.089	2.573	6.96	6.60	0.36

- a. S(CH₁₄) is the E-state value for the first carbon in the substituent R₆.
 b. S(NH₁₁) is the E-state value for the NH adjacent to the carbonyl group of the hydrazide functional group.
 c. pIC₅₀ is the negative log of the inhibitory power IC₅₀(μM).
 d. Calc is the calculated activity (pIC₅₀) from Eq. 8.
 e. Res. = pIC₅₀ - Calc.

What to assign as E-state value of the atom type not present?

- E-state value of '0' is valid result so reporting value of '0' for missing atom type should not be used (as in C2 – Accelrys)
- Use of -999 as E-state value for missing atom types as input for QSAR

What are the issues with E-states and multi-functional molecules?

- 35 atom types that are the bases for calculating K-H E-states are too general
- When dealing with QSAR for datasets where atom-by-atom matching is not possible and any given atom type hit more than once → the result is ambiguity that no statistical tool will resolve

More on ambiguity

- For example: ssNH could be part of
 - Sulphonamide, RNHSO₂R and
 - Amine, RNHR
 - Same atom type, both part of the same molecule – but in very different chemical environment
- What to calculate?
 - An average
 - Sum or
 - Both – the sum and the average

Testing hypothesis that simple counts should do at least as good as information rich K-H E-states

- Develop the program that will read in the same atom types and do the counts
- Choose several datasets that from QSAR area that feature multi functional type of molecules
- Use the same statistical approach to compare the performance of two sets of descriptors

Protocol used for comparison

- Descriptors:
 - E-state
 - 35 descriptors based on average E-state values
 - 35 descriptors based on sum of E-states
 - Counts
 - 35 based on the counts of K-H –state atom types
- Datasets
 - logP*, aqueous solubility, Human Intestinal Absorption and Blood Brain Barrier
- Statistical Tools
 - PCA/PLS in Simca (Umetrics)

Smarts Definitions for Kier-Hall Atom Types

RowNo	smarts-definitions	estates-atom-types-KH	RowNo	smarts-definitions	estates-atom-types-KH
1	[OH1][*]	sOH	19	S(=[*])(=[*])([*])[*]	ddssS
2	O=[*]	dO	20	[F][*]	sF
3	[OH0]([*])[*]	ssO	21	[Cl][*]	sCl
4	[o]	aaO	22	[Br][*]	sBr
5	[NH2][*]	sNH2	23	[I][*]	sl
6	[NH1]=[*]	dNH	24	[CH3][*]	sCH3
7	[NH1]([*])[*]	ssNH	25	[CH2]([*])[*]	ssCH2
8	[nH1]	aaNH	26	[CH2]=[*]	dCH2
9	N#[*]	tN	27	[CH1]([*])([*])[*]	sssCH1
10	[ND2](=[*])[*]	dsN	28	[CH1](=[*])[*]	dsCH1
11	[nH0]	aaN	29	[CH1]#[*]	tCH
12	N([*])([*])[*]	sssN	30	[cH]	aaCH
13	N(=[*])(=[*])[*]	ddsN	31	[cH0]	aaC
14	[N;+]([*])([*])([*])[*]	ssssN+	32	C(=[*])=[*]	ddC
15	[SH1][*]	sSH	33	C#[*][*]	tsC
16	S=[*]	dS	34	C(=[*])([*])[*]	dssC
17	[SX2]([*])[*]	ssS	35	C([*])([*])([*])[*]	ssssC
18	[s]	aaS			

Calculating E-state Descriptors

Name	%F (HIA)	sOH-sum	sOH-av	dO-sum	dO-av	ssO-sum	ssO-av	aaO-sum
raffinose	0.3	108.94	9.9	-999	-999	26.46	5.29	-999
lactulose	0.6	76.52	9.56	-999	-999	15.31	5.1	-999
aztreonam	1	18.06	9.03	57.84	11.57	4.92	4.92	-999
ceftriaxone	1	9.89	9.89	62.29	12.46	4.74	4.74	-999
cefuroxime	1	9.5	9.5	47.57	11.89	9.3	4.65	5.13
kanamycin	1	70.91	10.13	-999	-999	22.2	5.55	-999

Counts of Kier-Hall Atom Types

Name	%F (HIA)	sOH	dO	ssO	aaO	sNH2	dNH	ssNH	aaNH
raffinose	0.3	11	0	5	0	0	0	0	0
lactulose	0.6	8	0	3	0	0	0	0	0
aztreonam	1	2	5	1	0	1	0	1	0
ceftriaxone	1	1	5	1	0	1	0	1	1
cefuroxime	1	1	4	2	1	1	0	1	0
kanamycin	1	7	0	4	0	4	0	0	0

Objectives

- Compare quality of the models (R^2), based on training set alone and using in-built cross-validation Q^2 (LMO) within Simca
- Each of the datasets used has been analysed in the literature using similar approaches but with different descriptors
- NOT designed to build best models for those datasets

Performance of E-states vs Counts using Simca and PLS

e-states (ES)	counts of ES at-type	Performance	
R^2	R^2	$(R^2(\text{ES}) - R^2(\text{Counts})) * 100$	
0.655	0.659	-0.4	
0.306	0.49	-18.4	
0.611	0.59	2.1	
0.42	0.718	-29.8	

Conclusions

- Simple counts of the same atom types that Kier-Hall Estate descriptors are built on work at least as good in building the models for BBB and solubility, and outperform E-states when building models for HIA and logP, 18% and 30% respectively
- Reviewing recently submitted paper on modelling aqueous solubility, authors made the following observation:
 - Replacing E-states values by binary presentation of the K-H atom types, 1 if present and 0 if not did make much difference in model performance

Acknowledgment

- Thanks to Daylight for supplying programming toolkits for coding E-states algorithm and development of software for counting atom types based on smarts definitions